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Key indicators

Single-crystal X-ray study

T = 273 K

Mean $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$

R factor = 0.040

wR factor = 0.115

Data-to-parameter ratio = 13.7

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.*N'*-[(1*E*)-(5-Chloro-2-hydroxyphenyl)(phenyl)-methylene]-2-hydroxybenzohydrazide

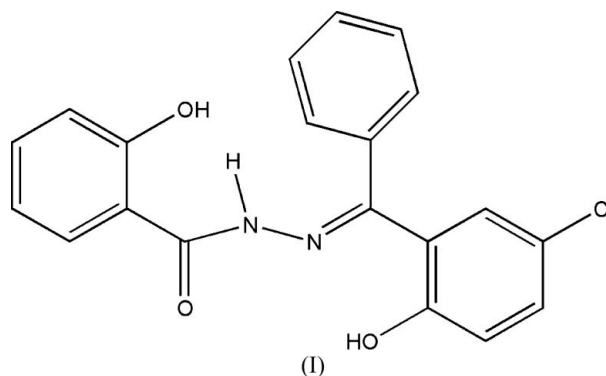
The title compound, $\text{C}_{20}\text{H}_{15}\text{ClN}_2\text{O}_3$, displays a *trans* configuration with respect to the $\text{C}=\text{N}$ double bond. The crystal structure is stabilized by intramolecular $\text{O}-\text{H}\cdots\text{N}$ and $\text{N}-\text{H}\cdots\text{O}$ and intermolecular $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonds.

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Comment

The chemistry of aroylhydrazones continues to attract much attention due to their ability to coordinate to metal ions (Singh *et al.*, 1982; Salem, 1998) and their biological activity (Singh *et al.*, 1982; Carcelli *et al.*, 1995). As an extension of our work on the structural characterization of aroylhydrazone derivatives, the title compound, (I), was synthesized and characterized.



The molecule of (I) displays a *trans* configuration with respect to the $\text{C7}=\text{N1}$ double bond (Fig. 1). The three benzene rings, $\text{C1}-\text{C6}$ (A), $\text{C8}-\text{C13}$ (B) and $\text{C15}-\text{C20}$ (C), make dihedral angles of $80.05(7)^\circ$ (A/B), $6.16(13)^\circ$ (A/C) and $81.95(7)^\circ$ (B/C).

Intramolecular $\text{N}-\text{H}\cdots\text{O}$ and $\text{O}-\text{H}\cdots\text{N}$ hydrogen bonds stabilize the conformation of the molecule, while $\text{O}-\text{H}\cdots\text{O}$ intermolecular hydrogen bonds lead to the formation of infinite chains parallel to the *b* axis (Table 1, Fig. 2).

Experimental

2-Hydroxybenzoylhydrazine (0.01 mol, 1.52 g) was dissolved in anhydrous ethanol (50 ml) and (5-chloro-2-hydroxyphenyl)-(phenyl)methanone (0.01 mol, 2.32 g) was added. The reaction mixture was refluxed for 8 h with stirring, and then the resulting precipitate was collected by filtration, washed several times with ethanol and dried *in vacuo* (yield 78%). The compound (1.0 mmol, 0.37 g) was dissolved in dimethylformamide (15 ml) and kept at room temperature for 25 d to obtain colourless single crystals suitable for X-ray diffraction.

Crystal data

 $C_{20}H_{15}ClN_2O_3$ $M_r = 366.79$ Orthorhombic, *Pbca* $a = 15.3237 (6) \text{ \AA}$ $b = 12.0171 (5) \text{ \AA}$ $c = 20.0773 (8) \text{ \AA}$ $V = 3697.2 (3) \text{ \AA}^3$ $Z = 8$ $D_x = 1.318 \text{ Mg m}^{-3}$ Mo $K\alpha$ radiation $\mu = 0.23 \text{ mm}^{-1}$ $T = 273 (2) \text{ K}$

Block, colourless

 $0.45 \times 0.24 \times 0.20 \text{ mm}$

Data collection

Bruker SMART CCD area-detector diffractometer

 φ and ω scans

Absorption correction: multi-scan

(SADABS; Sheldrick, 1996)

 $T_{\min} = 0.924$, $T_{\max} = 0.978$

39233 measured reflections

3248 independent reflections

2261 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.058$ $\theta_{\text{max}} = 25.0^\circ$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.040$ $wR(F^2) = 0.115$ $S = 1.00$

3248 reflections

237 parameters

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0507P)^2 + 1.2845P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\text{max}} = 0.001$ $\Delta\rho_{\text{max}} = 0.12 \text{ e \AA}^{-3}$ $\Delta\rho_{\text{min}} = -0.22 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|--------------------|-------|-------------|-------------|---------------|
| $O1-H1\cdots N1$ | 0.82 | 1.84 | 2.551 (1) | 145 |
| $N2-H2\cdots O3$ | 0.86 | 1.90 | 2.567 (1) | 133 |
| $O3-H3\cdots O2^1$ | 0.82 | 1.77 | 2.578 (1) | 169 |

Symmetry code: (i) $-x + \frac{3}{2}, y - \frac{1}{2}, z$.

All H atoms were positioned geometrically and treated as riding on their parent atoms, with $C-H(\text{aromatic}) = 0.93$, $O-H = 0.82$ and $N-H = 0.86 \text{ \AA}$, and with $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{O})$ or $1.2U_{\text{eq}}(\text{C}_{\text{aromatic}}, \text{N})$.

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997a); molecular graphics: SHELXTL (Sheldrick, 1997b) and PLATON (Spek, 2003); software used to prepare material for publication: SHELXTL.

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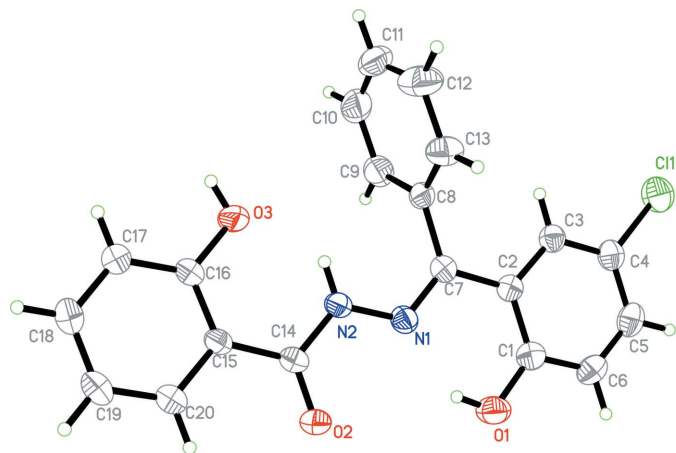


Figure 1

The molecular structure of compound (I), showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level and H atoms are represented as spheres of arbitrary radii.

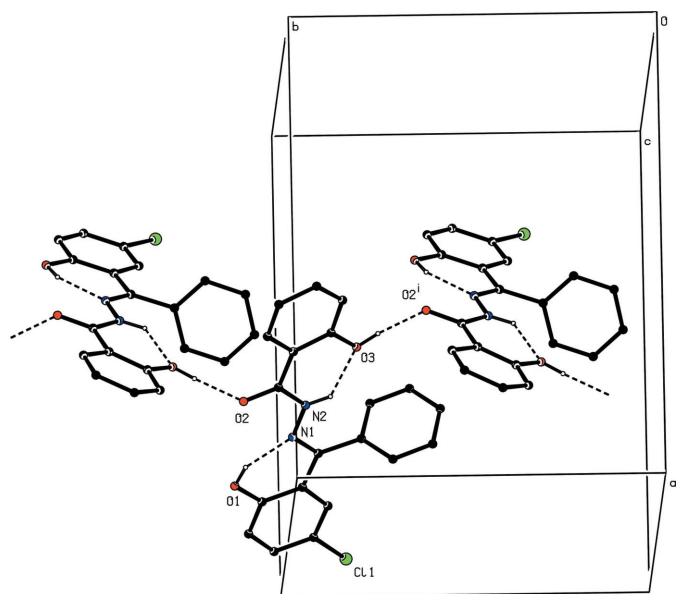


Figure 2

A partial packing view of (I), showing the formation of chains. Dashed lines show intra- and intermolecular hydrogen bonds. H atoms not involved in hydrogen bonds have been omitted for clarity. [Symmetry code: (i) $\frac{3}{2} - x, y - \frac{1}{2}, z$].

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